# Spectroscopy of the Heaviest Elements

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# Outline

### **Heavy Elements**

- What are they Why are they interesting
- **Heavy Element Detection**
- Spectroscopy Techniques
  - In-beam spectroscopy
  - Decay spectroscopy
  - K-isomers
  - SHE spectroscopy

### What is the atomic number of Flerovium?

- a) 112
- b) 114
- c) 116
- d) 118
- e) 120

# Upper End of Chart of Nuclides: 1966

- Predictions without shell effects:
  - Z=104 last element stable against SF



Myers and Swiatecki: Nucl. Phys. **81** (1966) 1 Patyk: Nucl. Phys. A **502** (1989) 591c

## What about the lifetimes?



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### $N \rightarrow$

Shell effects from Sobiczewski et al: Phys. Rev. C 63 (2001) 034306



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- Understanding of production and decay properties
- Determining the location of the next shells
- Evolution of shape
- Confirmation of assigned A and Z

Exotic Ecano sammer senser, magazet 2011

Shell effects from Sobiczewski et al: Phys. Rev. C 63 (2001) 034306

142

 $N \rightarrow$ 

144

146

148

90

136

138

140

# **Regions of Study**



- Understanding of production and decay properties
- Determining the location of the next shells
- Evolution of shape
- Confirmation of assigned A and Z

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## Single-Particle States Responsible for SHE



Deformation  $\varepsilon_2$ 

We can learn about the singleparticle states responsible for the stability of superheavy elements in two ways:

1) States which are near the ground state in deformed nuclei near Z=102 are also near the ground state for spherical SHE. We can study these states in Z=102-104 with production rates of up to atoms per second.

2) We can produce SHE at rates of nearly 1 atom per day. New capabilities allow us to perform spectroscopy directly with SHE.

Nilsson diagram for protons with Z≥82

### Spectroscopy of Heavy Elements - Challenges

#### Low cross sections:

Z~100: μb to nb
2~100: μb to nb
1 heavy element per 10<sup>12-14</sup> beam particles
and 10<sup>5-7</sup> unwanted reaction products
2~114: pb
1 heavy element per 10<sup>17</sup> beam particles
and 10<sup>11</sup> unwanted reaction products

Every aspect of spectroscopy dominated by background

### What you need for spectroscopy:

High-intensity beams (pµA) Way to tell when you actually make something interesting

### Two techniques recently developed:

In-beam spectroscopy

Decay spectroscopy

## Decay Spectroscopy – How

#### Technique to access Nuclear Structure in Heavy Element Isotopes

- 1) Produce heavy element
- 2) Separate and implant in focal plane detector
- 3) Observe decay in same detector pixel
- 4) Used known time of recoil and decay to gate on  $\gamma$ -rays



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## In-beam Spectroscopy - Why



Observing γ-rays that occur during the de-excitation process directly after a compound nucleus is formed

- Heavy element compound nuclei formed at coulomb barrier beam energies
- De-excite through combination of particle and γ-ray emission

• Process takes 
$$10^{-15}$$
 to  $10^{-13}$  s



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## In-beam Spectroscopy - Why



### **Rotational bands**

- Deformation
- how centrifugal force causes shape changes by stressing nuclei at high spin
- Single-particle states

Vibrational and octupole bands

All these can be used to put strong constraints on nuclear models, and improve our understanding of nuclei in this region at the high-Z limit of nuclear stabiltiy.

## Places with In-Beam Spectroscopy Setups





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## In-beam Spectroscopy – <sup>144</sup>Sm(<sup>36</sup>Ar,4n)<sup>176</sup>Hg



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## In-Beam Spectroscopy – What has been done



 $N \rightarrow$ 

Shell effects from Sobiczewski et al: Phys. Rev. C 63 (2001) 034306

# <sup>252</sup>No and <sup>254</sup>No from <sup>208,206</sup>Pb(<sup>48</sup>Ca,2n)



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# <sup>256</sup>Rf from <sup>208</sup>Pb(<sup>50</sup>Ti,2n)



Greenlees: Phys. Rev. Lett. 109 (2012) 012501

## In-beam Spectroscopy – What we have learned

### Deformation parameters for <sup>252,254</sup>No ≈ 0.29



## In-beam Spectroscopy – What we have learned

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Measured ground state rotational bands



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## In-beam Spectroscopy – What we have learned

Deformation parameters for  $^{252,254}$ No  $\approx 0.29$ 

Measured ground state rotational bands

Decay from states above the yrast line



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# In-beam Spectroscopy - Challenges

• Rate in Ge-detectors surrounding target limit beam intensity to ~20-50 pnA  $\rightarrow \sigma$ >20 nb



# Isotopes with Suitable Cross Sections

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# In-beam Spectroscopy - Challenges

- Rate in Ge-detectors surrounding target limits beam intensity to ~20-50 pnA  $\rightarrow \sigma$ >20 nb
- Isomers



Heavy elements are formed with 20-50 MeV of kinetic energy, or about 1-2% the speed of light. Given that most inbeam spectroscopy setups have radii of <30 cm, how long does it take a heavy element to exit the spectrometer?



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## Decay Spectroscopy – How



Technique to access Nuclear Structure in Heavy Element Isotopes

- 1) Produce heavy Element Isomers
- 2) Separate and implant in focal plane detector

## Decay Spectroscopy – How

New Technique to access Nuclear Structure in Heavy Element Isotopes

1) Produce heavy Element Isomers

2) Separate and implant in focal plane detector

3) Observe decay in same detector pixel

4) Observe  $\gamma$ -rays coincident with  $\alpha$ /c.e.



## Gammas



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## Decay Spectroscopy – Why



- Single-particle/multiquasiparticle states
- measurement of rotational, vibrationonal and octupole bands

# Investigation of K-isomers in Z~100

Nilsson diagram for protons with Z≥82

## K-Isomers – What

• K is the angular-momentum projection on the nuclear symmetry axis



## K-Isomers – What

- K is the angular-momentum projection on the nuclear symmetry axis
- K selection rules are similar to J selection rules: transitions with lower  $\Delta {\rm K}$  are favored



$$K=\Omega_{1+}\Omega_2$$

# K-Isomers – Where are they found



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# <sup>254</sup>No Isomers





A new generation of experiments is underway addressing the fundamental issue of the maximum limit of nuclear mass and charge.

## Going to higher Z – Spectroscopy of SHE



What is the difference between a  $\gamma$ -ray and an x-ray?

- a) X-rays are atomic transitions, γ-rays are nuclear transitions
- b) X-rays are nuclear transitions, γ-rays are atomic transitions
- c) X-rays are lower energy than γ-rays
- d) No difference

## What is needed for a Z identification



Dirac-Fock-Slater prediction of K x-ray energies – (B. Fricke, G. Soff, At. Data Nuc. Data Tab, **19**, 83 (1977)

Observation of 3 Kα X rays can provide unambiguous Z identification – in the absence of background



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## Spectrum of $\alpha$ -decays from E115 and daughters



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## γ-like events Coincident with '280111' Decays



Given the  $\alpha$  and  $\gamma$  spectra and ratios of emitted conversion electrons to  $\gamma$ -rays below, can we determine the multipolarity of the 237-keV transition? If so, what is it?



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## γ-like events Coincident with '276109' Decays



## Summary

In-Beam Spectroscopy has been performed on Fm, No and Rf isotopes

Ground state rotational bands Deformation

Hints of higher lying states

### **K-Isomer studies**

Ground state rotational bands

Where K-isomers exist

Higher lying states

E115 Spectroscopy

one excited state each in <sup>280</sup>111 and <sup>276</sup>109



### BERKELEY, CALIFORNIA

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